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## WESTIELLAMIDE, A BISTRATAMIDE-RELATED CYCLIC PEPTIDE FROM THE BLUE-GREEN ALGA WESTIELLOPSIS PROLIFICA

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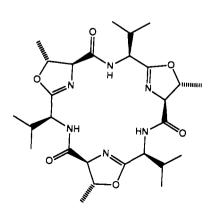
ABSTRACT.—The isolation and structural elucidation of westiellamide [1] from the terrestrial blue-green alga Westiellopsis prolifica is described. This moderately cytotoxic cyclic peptide appears to be identical with a bistratamide-type marine natural product from the aplousobranch ascidian Lissoclinum bistratum.

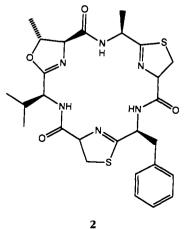
In aplousobranch ascidians belonging to the genus Lissoclinum, cyclic peptides such as patellamides and lissoclinamides from Lissoclinum patella (1) and bistratamides from Lissoclinum bistratum (2) appear to be located in the cells of the obligate prokaryotic algal symbionts (Prochloron spp.), strongly suggesting that the symbiotic prochlorophytes in the tunicates are involved in the biosynthesis of these secondary metabolites. Prochlorophytes resemble cyanophytes (blue-green algae) but differ in that they lack accessory phycobiliprotein pigments and possess both chlorophylls a and b (3). We report here the isolation and structural elucidation of westiellamide [1] from a terrestrial cyanophyte Westiellopsis prolifica Janet (Stigonematales, Stigonemataceae). Compound 1 is a moderately cytotoxic cyclic peptide which appears to be identical with a bistratamide-type compound that Hawkins et al. (4) have isolated from L. bistratum

and inappropriately named trisoxazoline in a recent patent.

Trisoxazoline is a generic term that had been used in an earlier unrelated patent (5). Westiellamide was cytotoxic against KB and LoVo cell lines at 2  $\mu g/$ ml, but was not solid tumor selective in the Corbett assay (6). Our finding of **1** in a cyanophyte provides additional circumstantial evidence that the *Prochloron* sp. in *L. bistratum* is responsible for the production of the bistratamides.

The blue-green alga was collected from a mud sample on the island of Oahu, Hawaii, and identified as W. prolifua. By repeated subculture, a clonal non-axenic culture of the cyanophyte, assigned strain number EN-3-1, was produced. Mass cultivation of the alga was carried out in a liquid medium (BG-11) using a general procedure that we have described elsewhere (7). The freezedried alga was extracted with 70% EtOH and the extract subjected to nor-





mal and reversed-phase chromatography to give westiellamide [1] in 0.06% yield.

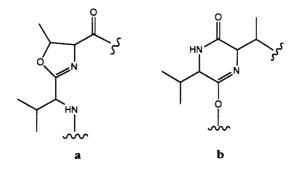
The structure of 1 was deduced by spectral analysis and chemical degradation. Hrms established the molecular formula  $C_{27}H_{42}N_6O_6$ . The nmr spectra, however, indicated that 1 was a trimer of a unit having the composition  $C_9H_{14}N_2O_2$ . Only nine signals could be observed in the <sup>13</sup>C-nmr spectrum: three methyl, four methine, and two quaternary carbon signals. The <sup>1</sup>H-nmr spectrum of 1 exhibited eight signals, seven of which correlated with the seven protonated-carbon signals in a HMOC experiment. The eighth signal was assigned to an exchangeable amide NH proton. <sup>1</sup>H-<sup>1</sup>H COSY, and <sup>1</sup>H-<sup>13</sup>C HMBC experiments strongly suggested that the repeating unit in 1 was a valulthreonul unit that had lost the elements of  $H_2O$ . We considered two structures **a** and **b** for this unit, but **a** fit the data better. The <sup>1</sup>H and <sup>13</sup>C chemical shifts and coupling constants were very similar to those reported for the same a unit in bistratamide A [2] (2) and patellamide C (8). Furthermore, mild acid hydrolysis (2 N HCl/MeOH, 2 h reflux) led to a complex mixture of products, one of which appeared to be cyclo[Thr-Val-Thr-Val-Thr-Val}, since a peak was observed at m/z 600 in the eims of the acid hydrolyzate. If westiellamide had been a trimer of **b**, we would have expected the imidate linkages to cleave readily on even milder acid hydrolysis to give 3-(1hydroxyethyl)-6-isopropyl-2,5-piperazinedione as the major product. Westiellamide was therefore a trimer of  $\mathbf{a}$ . Complete acid hydrolysis of westiellamide led to L-valine and L-threonine, which established the absolute stereochemistry as depicted in  $\mathbf{1}$ .

### **EXPERIMENTAL**

GENERAL PROCEDURES.—<sup>1</sup>H- and <sup>13</sup>C-nmr chemical shifts are referenced to solvent peaks:  $\delta_H$ 7.26 (residual CHCl<sub>3</sub>) and  $\delta_C$  77.0 for CDCl<sub>3</sub>. Homonuclear <sup>1</sup>H connectivities were determined with the COSY experiment, and heteronuclear <sup>1</sup>H-<sup>13</sup>C connectivities were determined by HMQC (9) and HMBC (10) experiments.

CULTURE CONDITIONS.—A clonal isolate, designated UH strain EN-3-1 and identified as *W. prolifica*, was obtained from a freshwater mud sample on the island of Oahu, Hawaii. Repeated subculture on a solidified medium was used to purify the alga. Unialgal, nonaxenic cultures of EN-3-1 were grown on BG-11 medium in 20liter glass bottles as described elsewhere (7). After 24 to 30 days, the alga was harvested by filtration and freeze-dried. Yields of lyophilized alga averaged 0.098 g/liter.

ISOLATION.—The freeze-dried alga (17.3 g) was macerated in a blender and extracted with 70% EtOH (700 ml) three times. The combined extract was filtered and the solvent removed in vacuo. The crude extract (3.5 g) was fractionated by reversed-phase flash chromatography on a column of C18 silica (YMC gel ODS 120 Å, 50 g), using a steep-stepped gradient from H<sub>2</sub>O to MeOH to CH2Cl2. Repeated cc on Si gel (Davisil 200-425 mesh, 60 Å) using a hexane/EtOAc gradient, followed by further reversed-phase flash column chromatography with an H<sub>2</sub>O/MeOH gradient, yielded pure westiellamide [1] (9.6 mg) as an amorphous solid:  $[\alpha]D + 130^{\circ}$  (MeOH, c = 0.1; <sup>1</sup>H nmr (500 MHz, CDCl<sub>3</sub>)  $\delta$  7.77 (d, J = 7.5 Hz, N-H), 4.81 (m, J = 8.6, 5.9 Hz, H-3), 4.65 (dt, J = 7.5 and 2.7 Hz, H-6), 4.22 (dd, J = 8.6 and 2.2 Hz, H-2), 2.29 (m, J = 7.0, 6.5,



2.7 Hz, H-7), 1.58 (d, J = 5.9 Hz, H-4), 0.89 (d, J = 7.0 Hz, H<sub>3</sub>-8), 0.82 (d, J = 6.5 Hz, H<sub>3</sub>-9); <sup>13</sup>C nmr (125 MHz, CDCl<sub>3</sub>) δ 170.54 (C-1), 168.46 (C-5), 82.66 (C-3), 73.76 (C-2), 52.32 (C-6), 31.43 (C-7), 21.92 (C-4), 18.60 (C-8), 16.83 (C-9); eims m/z (rel. int.) [M]<sup>+</sup> 546 (43),  $[M - Me]^+$  531 (5),  $[M - C_3H_7]^+$  503 (43),  $[M - C_4 H_8]^+$  490 (28),  $[M - C_4 H_8 O - CO]^+$ 462 (41), 363 (7), 308 (12), 280 (22), 252 (10), 169 (35), 138 (50), 69 (100); hreims m/z546.3182 (calcd for C<sub>27</sub>H<sub>42</sub>N<sub>6</sub>O<sub>6</sub>, -1.6 mmu error), 531.2959 (calcd for C26H39N6O6, -2.8 mmu error), 308.1615 (calcd for C15H22N3O4, -0.5 mmu error), 280.1663 (calcd for C14H22N3O3, -0.2 mmu error), 252.1711 (calcd for  $C_{13}H_{22}N_3O_2$ , +0.1 mmu error).

HYDROLYSIS AND AMINO ACID ANALYSIS.-Westiellamide (2 mg) in 6 N methanolic HCl was heated at 110° for 36 h. The amino acid hydrolyzate (2 mg) was heated with acetyl chloride (1.25 ml) and iPrOH (5 ml) at 100° for 45 min. The mixture was evaporated to dryness, and the residue was treated with pentafluoropropionic anhydride (2 ml) in CH<sub>2</sub>Cl<sub>2</sub> (3 ml) at 100° for 15 min. After cooling, excess reagent was evaporated with a stream of  $N_2$ . The mixture of derivatized amino acids in MeOH was analyzed by gc-ms using an Alltech Chirasil-Val column (25  $m \times 0.25$  mm) and the following conditions: column temperature 60→110° at 2°/min and a 12 psi head pressure (flow rate estimated to be about 0.6 ml/min). Using selective ion monitoring, the retention times for the N-pentafluoropropionyl isopropyl ester (PFP-IPA) derivatives of the amino acids from the westiellamide hydrolyzate were found to be 14.37 (L-valine) and 14.68 min (L-threonine). Retention times for the PFP-IPA derivatives of authentic amino acids were found to be 12.85 (D-valine), 13.52 (D-threonine), 14.27 (L-valine), 14.52 (L-threonine), 19.77 (D-allothreonine), and 21.13 min (L-allo-threonine).

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NOTE ADDED IN PROOF: The Hawkins group has renamed 1 from *L. bistratum* cycloxazoline. T.W. Hambley, C.J. Hawkins, M.F. Lavin, A. van der Brenk, and D.J. Watters, *Tetrahedron*, in press.